

MYP Biology

Ecosystems

Ecosystems

I.

Main Idea: Physical laws control energy flow and chemical cycling.

Main Idea: Energy *flows through* ecosystems.

Main Idea: Chemicals (matter) *cycles within* ecosystems.



PHYSICAL LAWS GOVERN ENERGY FLOW & CHEMICAL CYCLING IN ECOSYSTEMS

- **2 Fundamental Processes of Ecosystems:**
 - Energy Flow and Chemical Cycling.

A. Conservation of Energy

- **The sun is the ultimate source of energy for most ecosystems and life itself on our planet.**
- Energy enters ecosystems as solar radiation.
- Autotrophs *transform* solar radiation into chemical energy.
- Heterotrophs consume autotrophs and *transfer* this chemical energy through food chains.
- **First Law of Thermodynamics:** energy can not be created nor destroyed...only transferred and transformed.

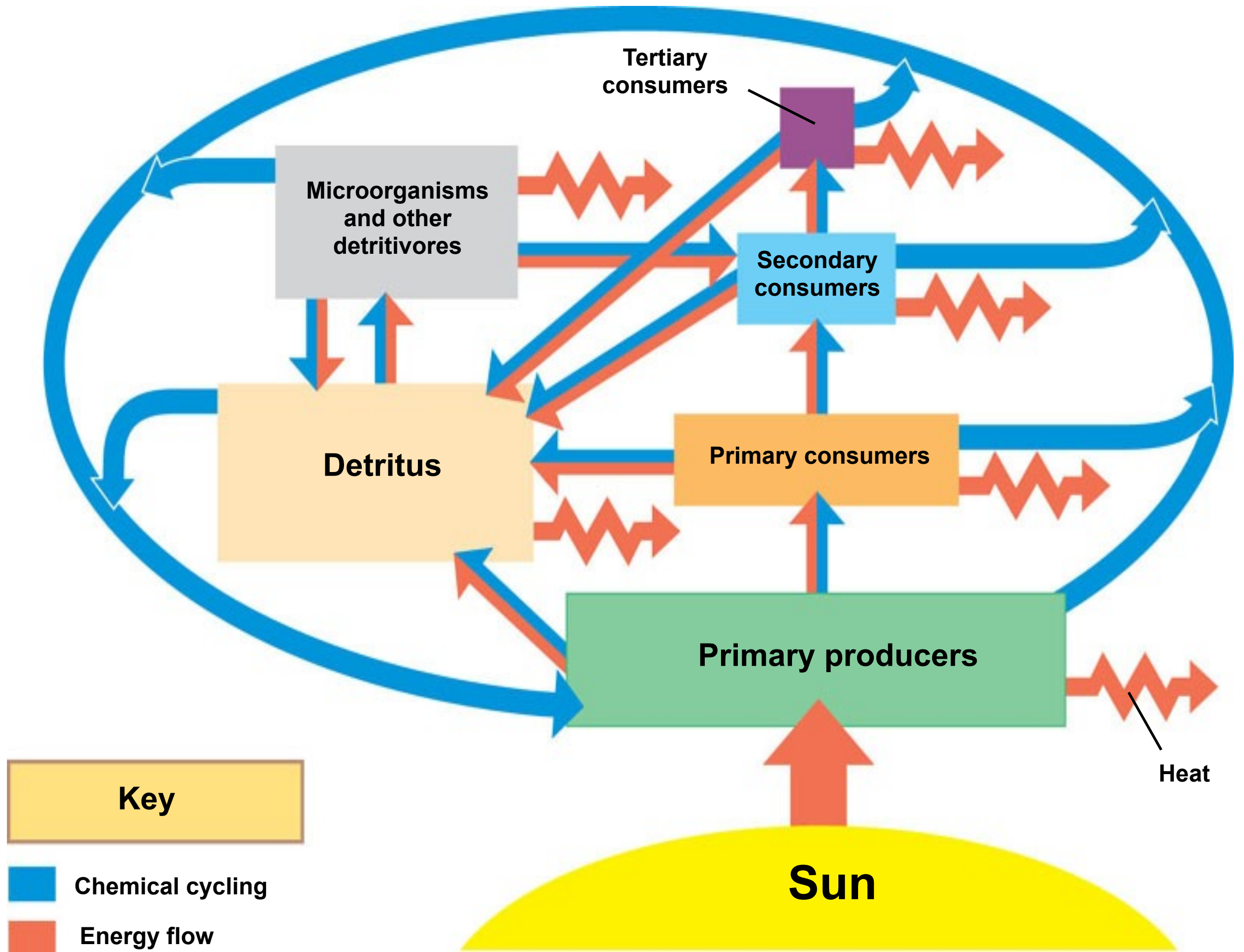
- **Second Law of Thermodynamics:** energy exchanges (transfers and transformations) increase the entropy of the universe. In other words energy exchanges are inefficient some energy is always lost as heat
- Remember energy is passing through food chains.
- As chemical energy is passing through food chains it is ultimately *transformed* to heat and *transferred* back to space

B. Conservation of Mass

- **Law of Conservation of Mass:** like energy, matter can not be created or destroyed.
- Very little matter enters earth from space.
- Thus the matter on *earth* is constant. Matter has been, is and will continue to be recycled over time.
- Be Careful! Matter can be gained and lost from *ecosystems* themselves.

C. Energy, Mass & Trophic Levels

- **Primary Producers** consists of *autotrophs* and their trophic level supports all others. Think of it this way...autotrophs link heterotrophs to the ultimate source of energy for most life...the sun!
- Producers (autotrophs) are *photosynthetic organisms* that use solar energy to synthesize organic compounds (sugars) which they use to fuel cellular respiration and as building blocks for growth.



- **Decomposers/Detrivores** consume **Detritus**, nonliving organic material (dead organisms and feces)
- Two most significant and important decomposers are Fungi and Bacteria
- They play a critical role in recycling matter...decomposers convert organic material into inorganic material that producers can then uptake and reuse. (recycled back into the ecosystem)
 - Consider This! If decomposition stopped detritus would build up and life would not exist



Ecosystems

II.

Main Idea: Energy and other limiting factors control primary production.

Main Idea: Primary production dictates the energy budget for the entire ecosystem.



ENERGY AND OTHER LIMITING FACTORS CONTROL PRIMARY PRODUCTION IN ECOSYSTEMS

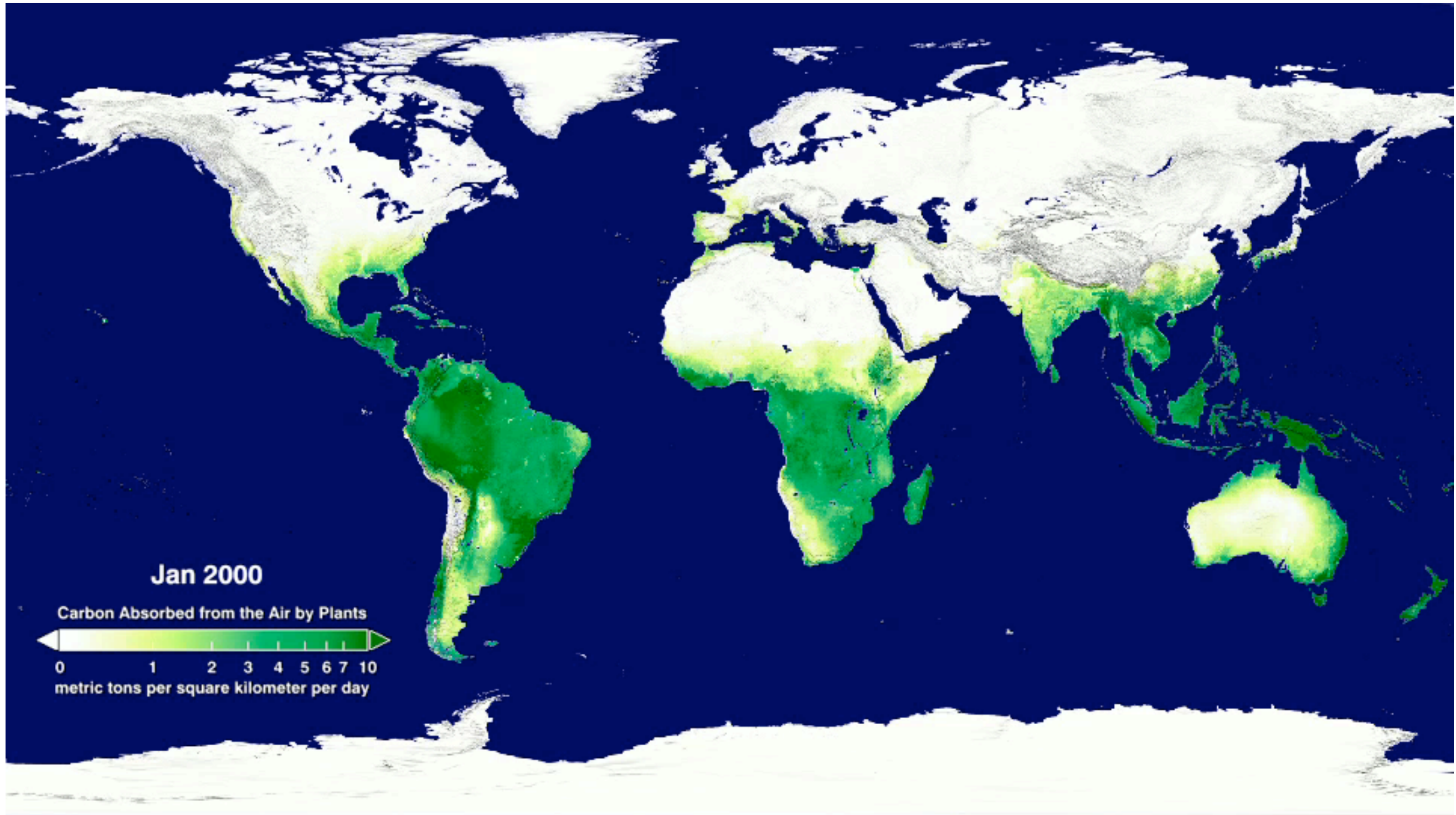
- **Primary production** is the amount of light energy converted to chemical energy in a given period of time
- The amount of light energy converted to chemical energy by autotrophs affects the amount of heterotrophs it can support in higher trophic levels.
- **Energy transfer is a major theme in biology, energy transfer underlies all biological interactions.**

A. Ecosystem Energy Budgets

I. Global Energy Budgets

- WOW...In one day the earth receives enough solar energy to power every human beings' energy need for the next 20+ years at our current consumption rate. BUT...
- A lot of solar radiation is absorbed, scattered or reflected by our atmosphere.
- Most of the solar radiation that does reach the earth contacts things that can not photosynthesize and of the light that reaches autotrophs only certain wavelengths power photosynthesis
 - All in all about 1% of visible light is converted to chemical energy.

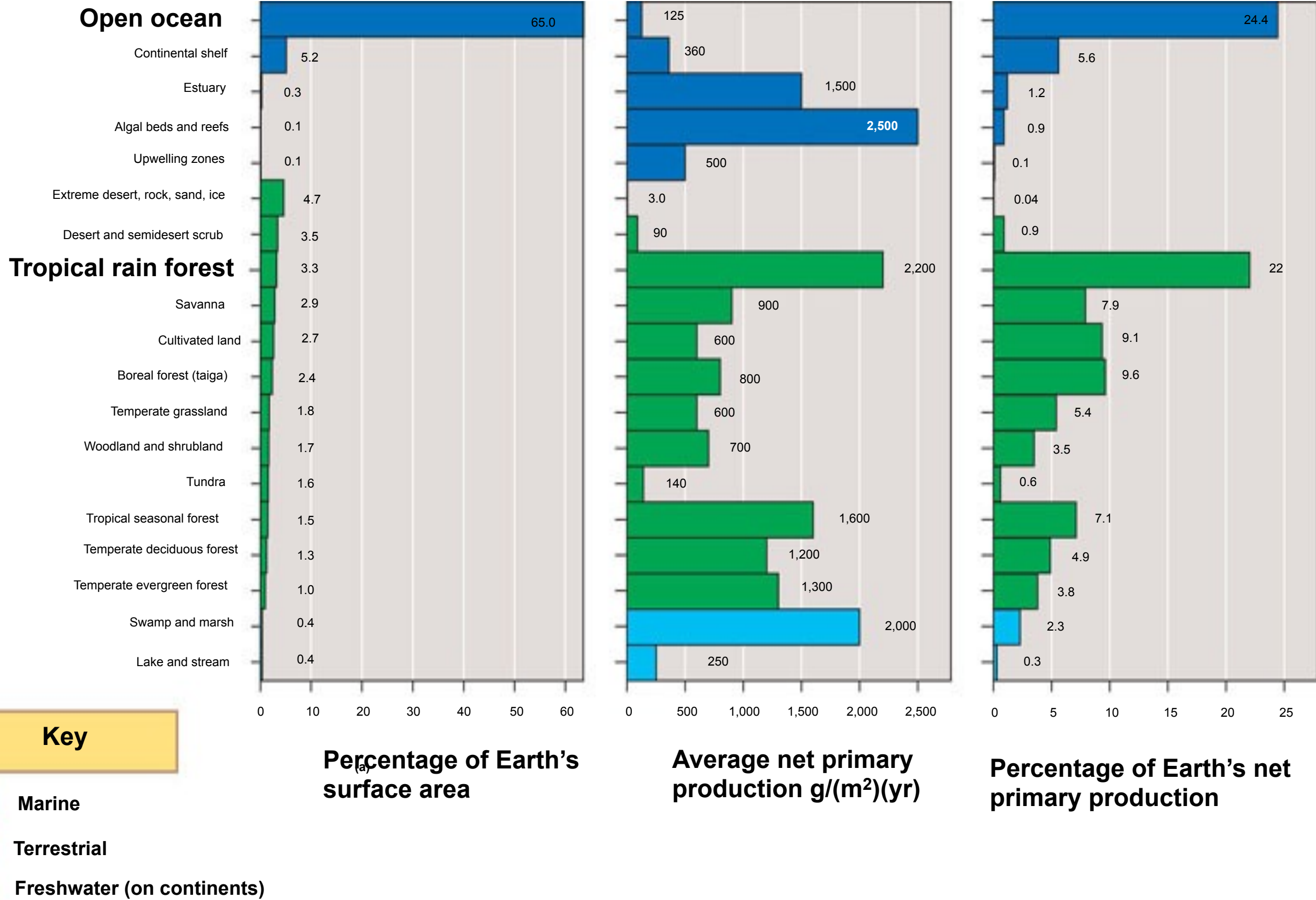
- **Gross Primary Production (GPP)** is the total amount of solar energy converted to chemical energy

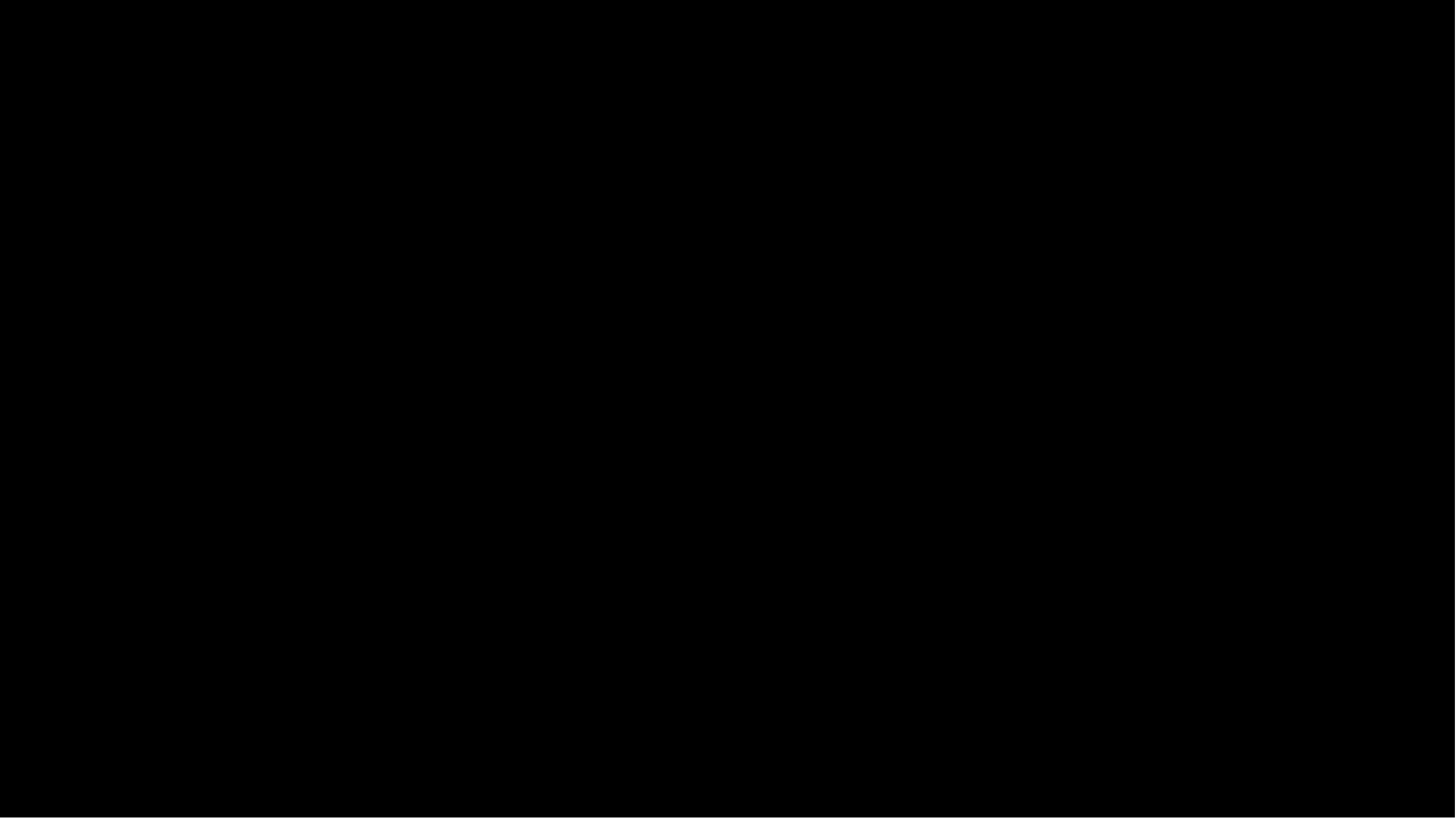


The gross primary productivity of the world's land areas for the period 2000-2009 as calculated from Terra's MODIS instrument. The original 8-day average GPP data has been smoothed to a 24-day average to make the animation less noisy.

2. Gross and Net Production

- **Gross Primary Production (GPP)** is the total amount of solar energy converted to chemical energy per unit time.
- **Net Primary Production (NPP)** is the difference between GPP and autotrophic respiration R_a .
- $NPP = GPP - R_a$
 - Think of R_a as the energy that plants use to “run themselves” whatever energy is leftover can be used by the plant to add new biomass (grow)
 - This new biomass contains the chemical energy for heterotrophs





The past decade is the warmest on record since instrumental measurements began in the 1880s. Previous research suggested that in the '80s and '90s, warmer global temperatures and higher levels of precipitation -- factors associated with climate change -- were generally good for plant productivity. An updated analysis published this week in Science indicates that as temperatures have continued to rise, the benefits to plants are now overwhelmed by longer and more frequent droughts. High-resolution data from the Moderate Resolution Imaging Spectroradiometer, or MODIS, indicate a net decrease in NPP from 2000-2009, as compared to the previous two decades.

B. Primary Production in Aquatic Ecosystems

I. Light Limitations

- Light is the obvious choice as limiting factor.
- After all more than half of the solar radiation is absorbed in the first 15 meters of water.
- *Light is very important however it does not appear to be the key limiting factor in aquatic ecosystems!*

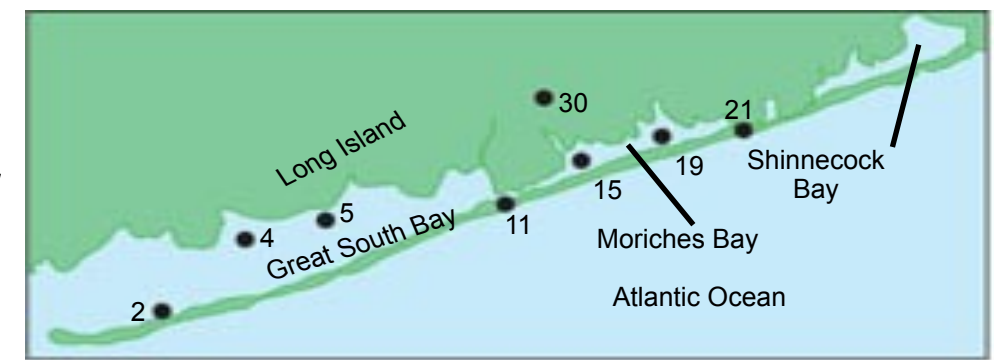
Assuming that light was the key limiting factor scientists predicted that a gradient in production from the poles to the equator. Would the production gradient increase or decrease as moved away from the poles? Why? Does this gradient exist?

2. Nutrient Limitations

- *Nutrients appear to be the key limiting factor in aquatic ecosystem production!*
- The most common limited nutrients in aquatic ecosystems are the (macronutrients) nitrogen and phosphorous.
- *Further support is found in upwellings (deep nutrient rich waters that circulate to the ocean surface), Areas of upwellings are diverse and abundant with life...they are often prime fishing locations.*

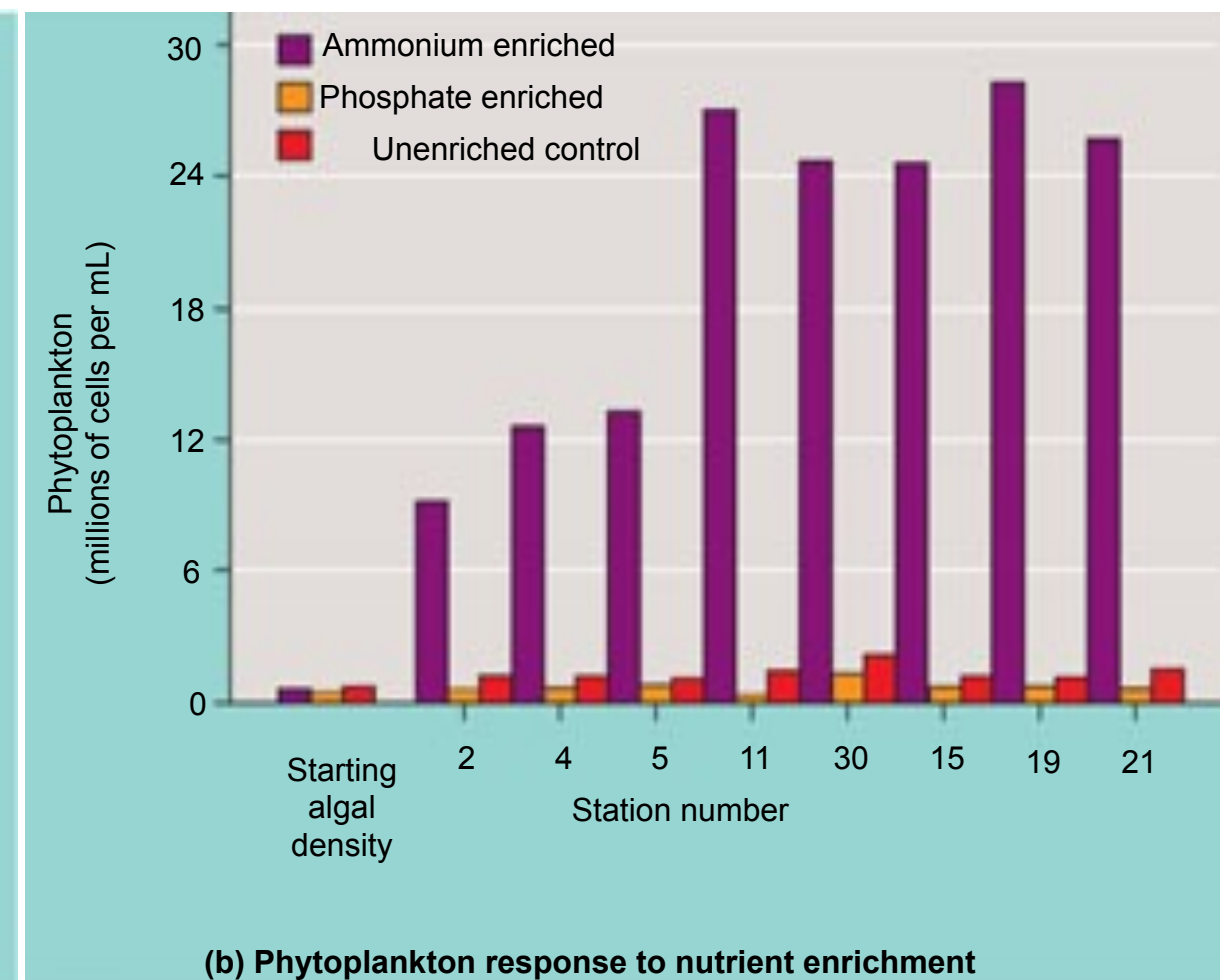
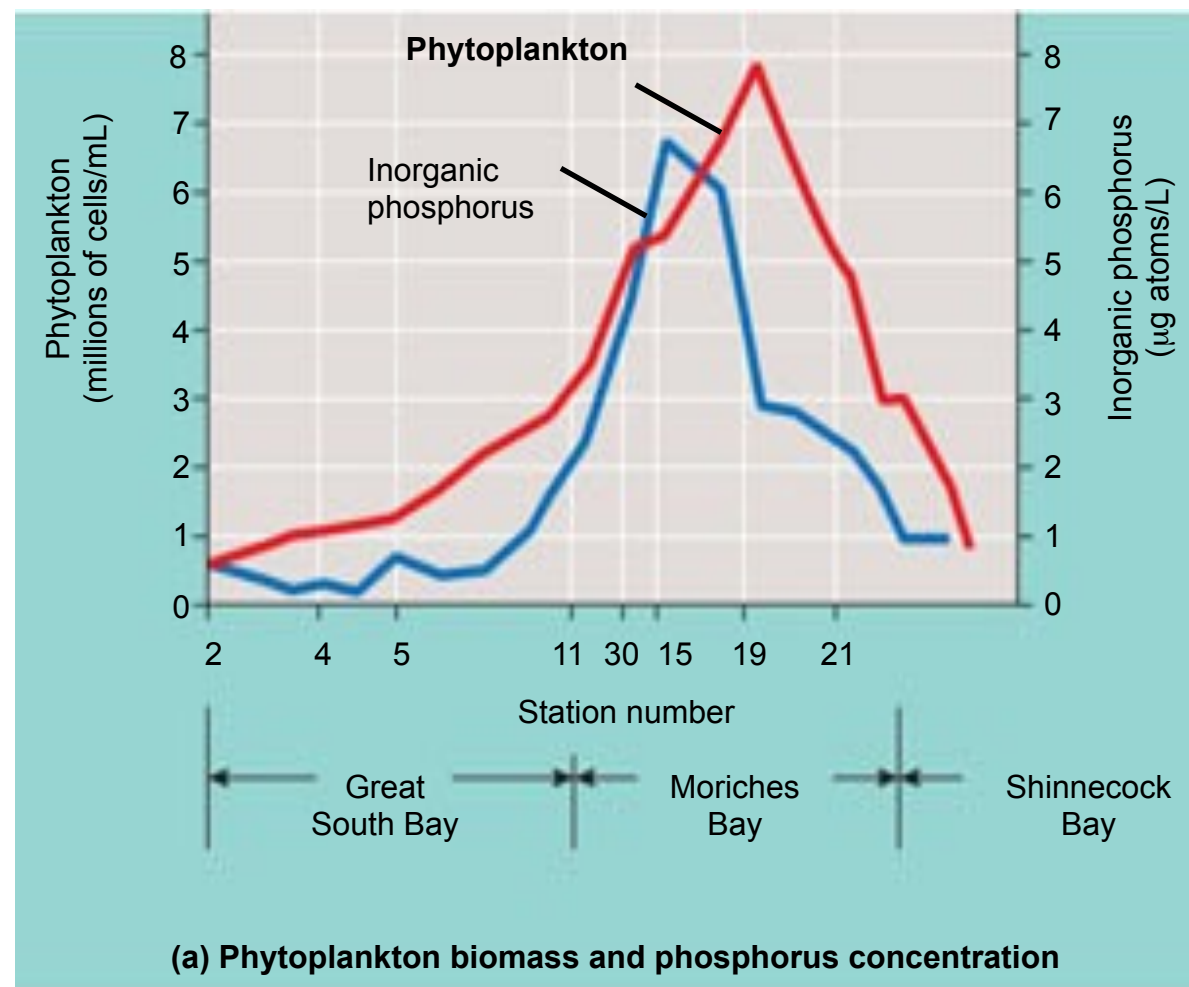
EXPERIMENT

Pollution from duck farms concentrated near Moriches Bay adds both nitrogen and phosphorus to the coastal water off Long Island. Researchers cultured the phytoplankton *Nannochloris atomus* with water collected from several bays.



RESULTS

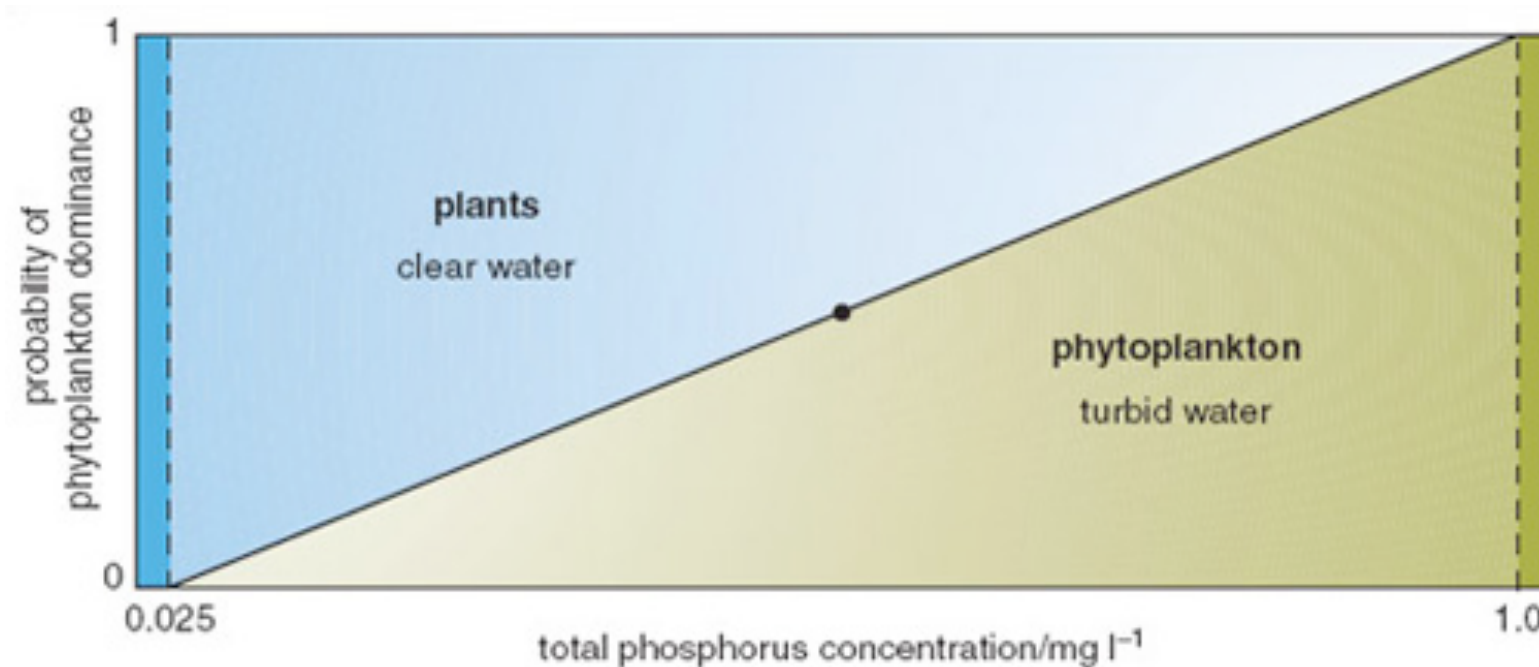
Phytoplankton abundance parallels the abundance of phosphorus in the water (a). Nitrogen, however, is immediately taken up by algae, and no free nitrogen is measured in the coastal waters. The addition of ammonium (NH_4^+) caused heavy phytoplankton growth in bay water, but the addition of phosphate (PO_4^{3+}) did not induce



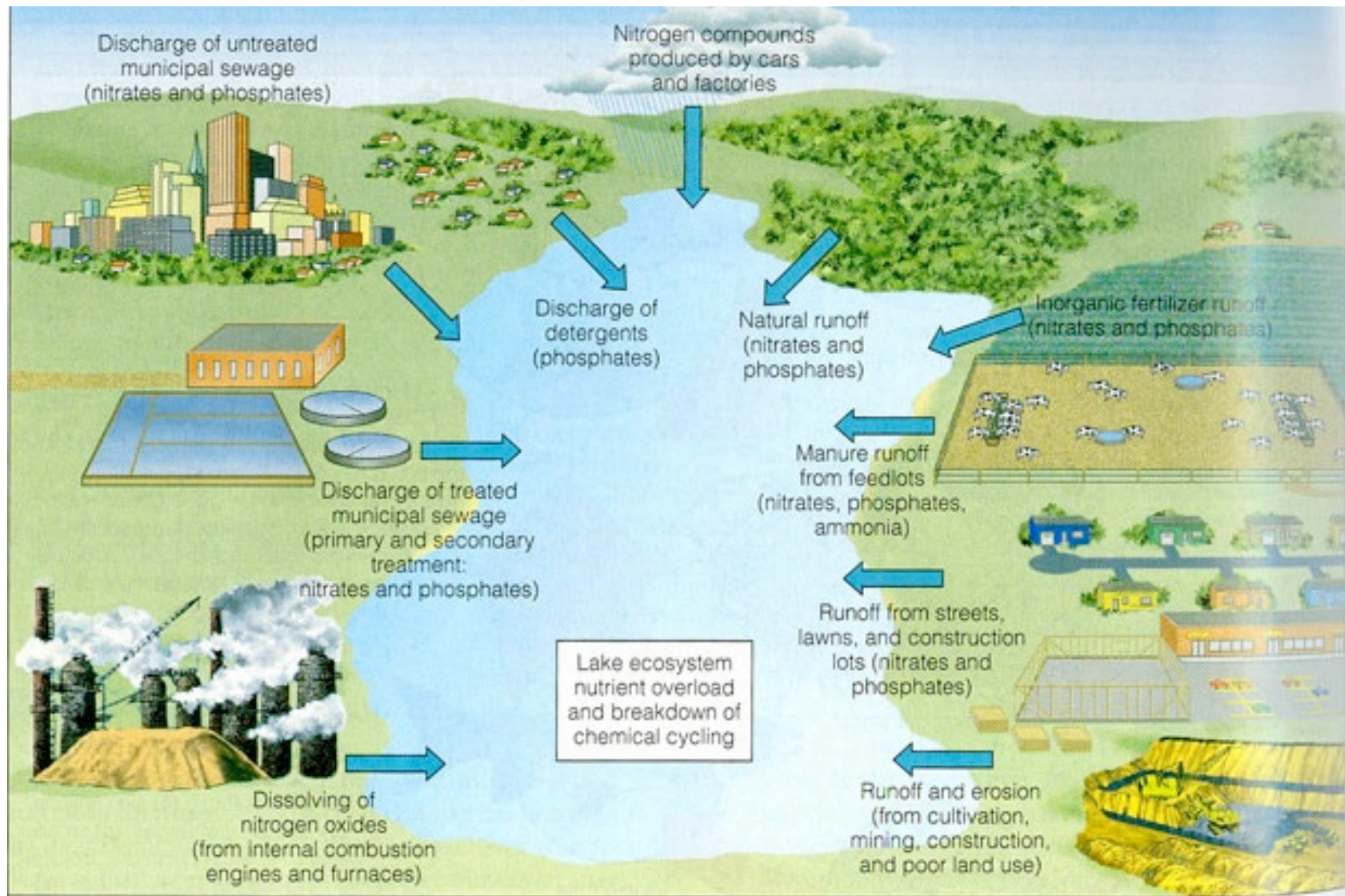
CONCLUSION

Since adding phosphorus, which was already in rich supply, had no effect on *Nannochloris* growth, whereas adding nitrogen increased algal density dramatically, researchers concluded that nitrogen was the nutrient limiting phytoplankton growth in this ecosystem.

Increase in Phosphorous can cause Eutrophication

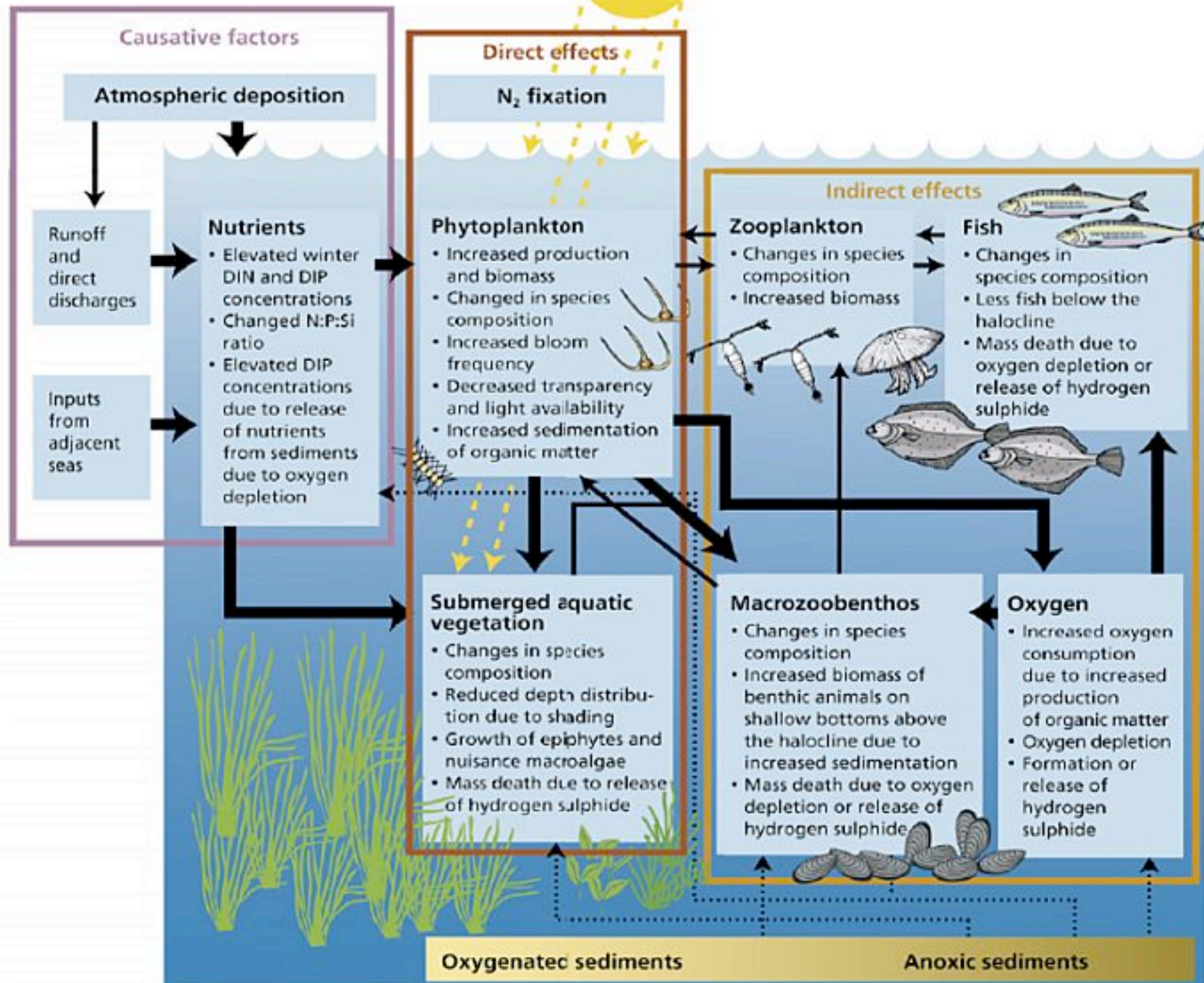


Cultural Eutrophication



Eutrophication:

Cause & Effect



C. Primary Production in Terrestrial Ecosystems

- **Water** and **Temperature** are the key limiting factors.
- In fact precipitation is a useful tool for predicting productivity because there is such a strong correlation between the two.

I. Adaptations to Nutrient Limitations

- Globally nitrogen is the most common limiting factor.
- *Various adaptations have evolved to address these challenges*
- *I. Root Nodules (symbiosis between prokaryotes and plants)*



Ecosystems

III.

Main Idea: Energy transfer is not efficient, most energy is lost.

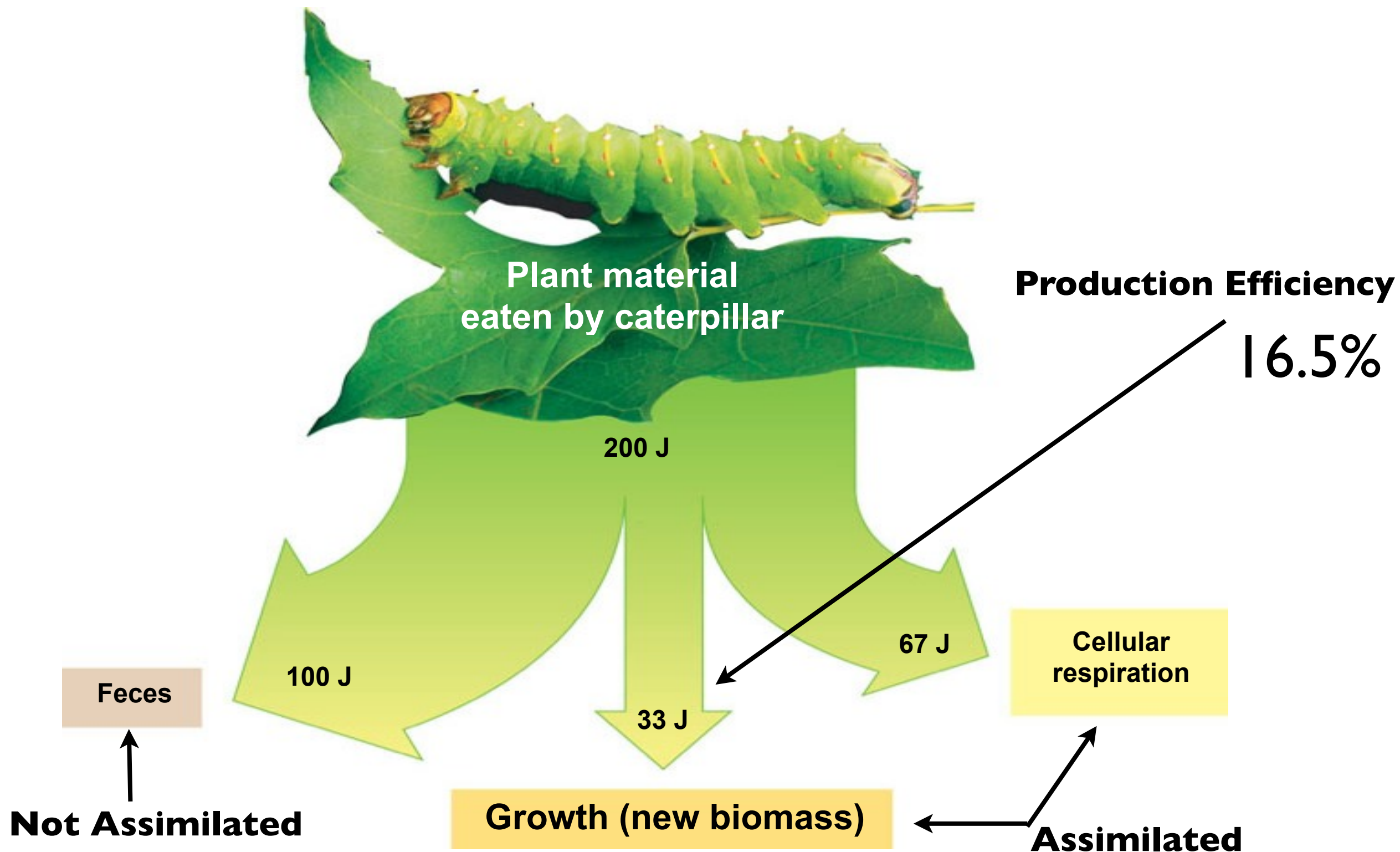


ENERGY TRANSFER BETWEEN TROPHIC LEVELS IS TYPICALLY AROUND 10% (WITH 90% LOST)

- **Secondary production:** the amount of chemical energy in consumers food that is actually converted to their own *new biomass* during some time period.

A. Production Efficiency

- **Think about it...** only the chemical energy stored by herbivores is in their biomass (either in their own biomass or the biomass of their offspring)
This is only energy available to secondary consumers!
- We can measure production efficiency.



Match the pictures with their production efficiencies.



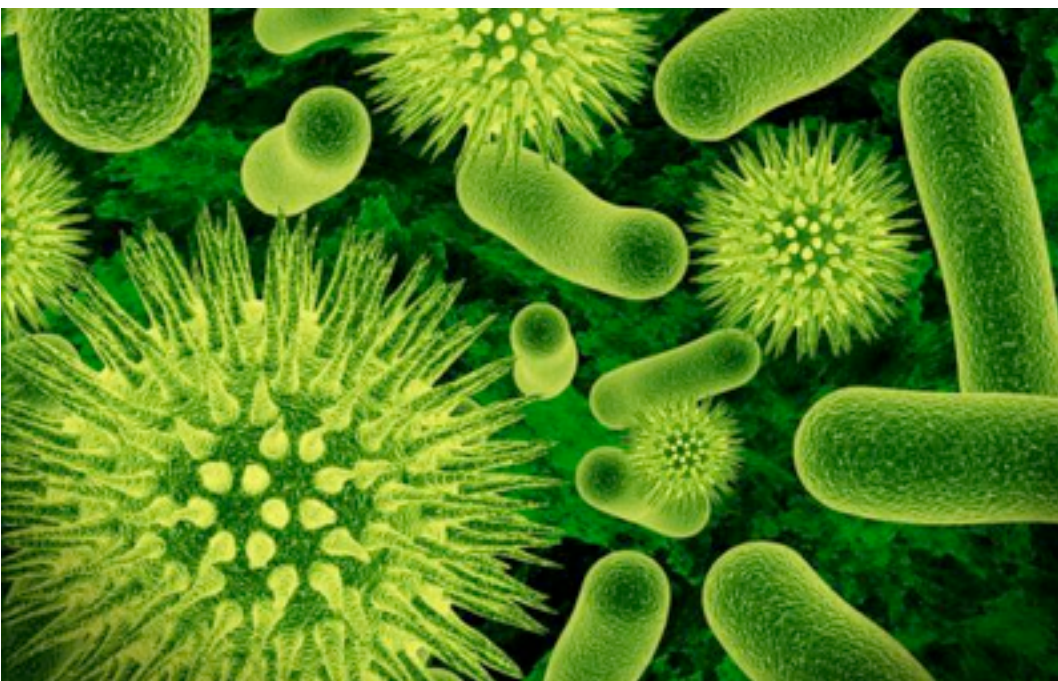
1-3%

1-3%

10%

40%

90%

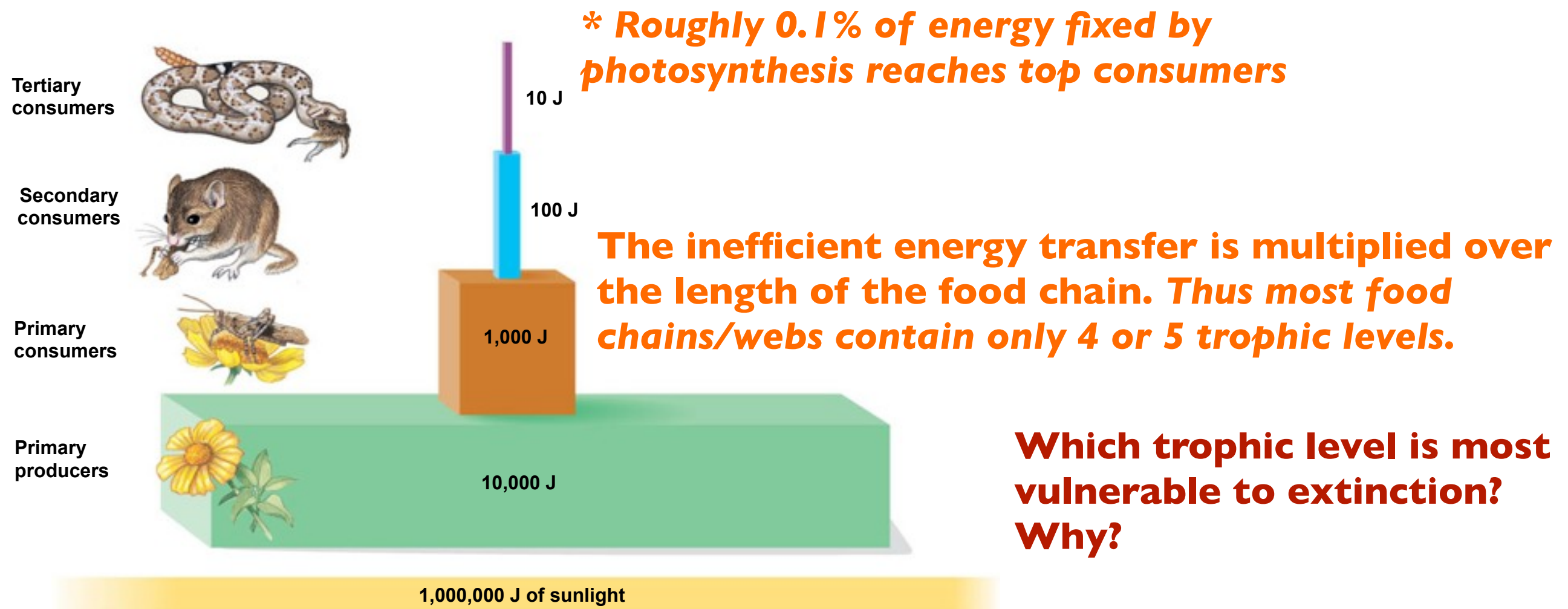


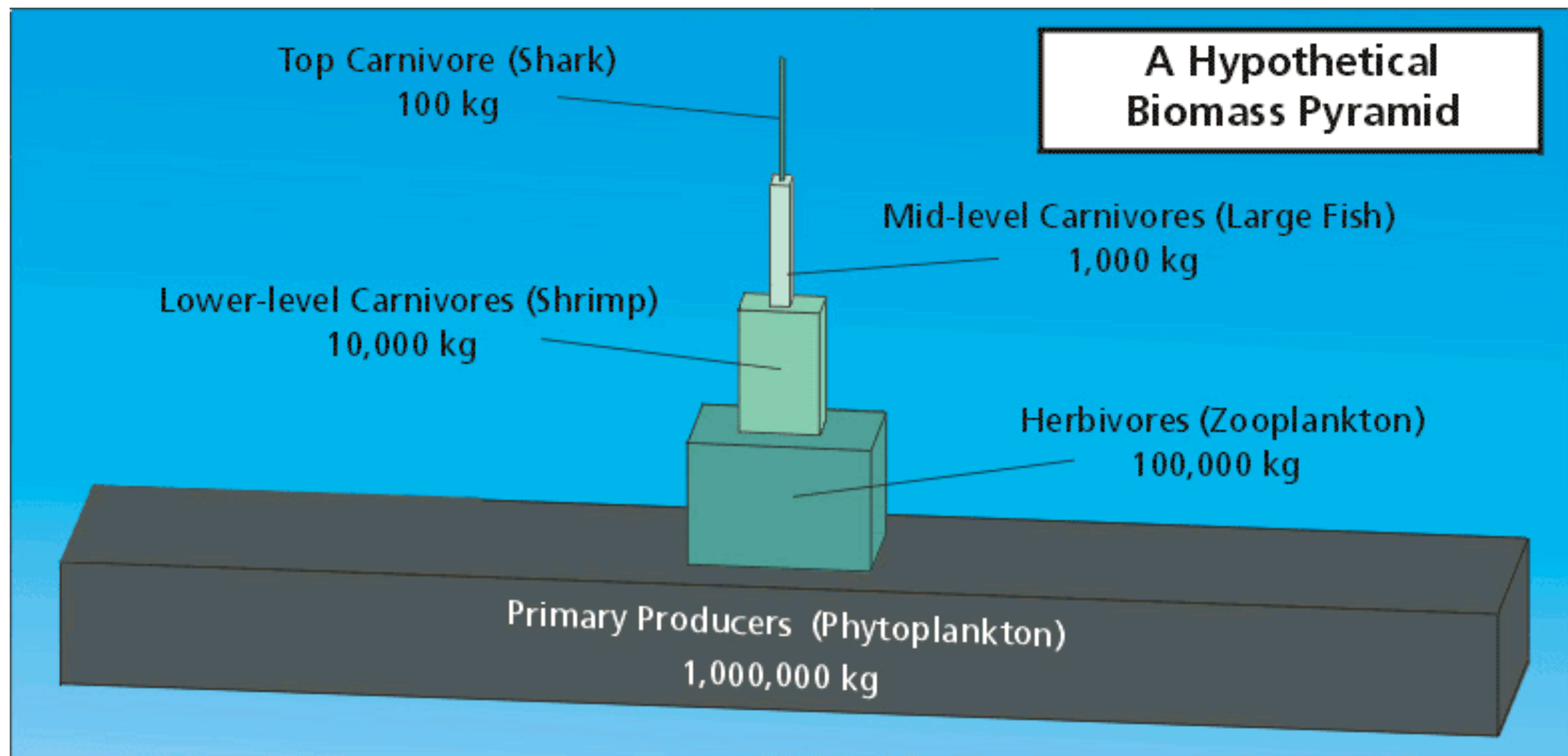
Explain
your
choices.

Birds & Mammals(1-3%), alot of energy maintaining body heat
Fish (10%)
Microorganisms (40%) due efficient surface area to volume ratio

B. Trophic Efficiency & Ecological Pyramids

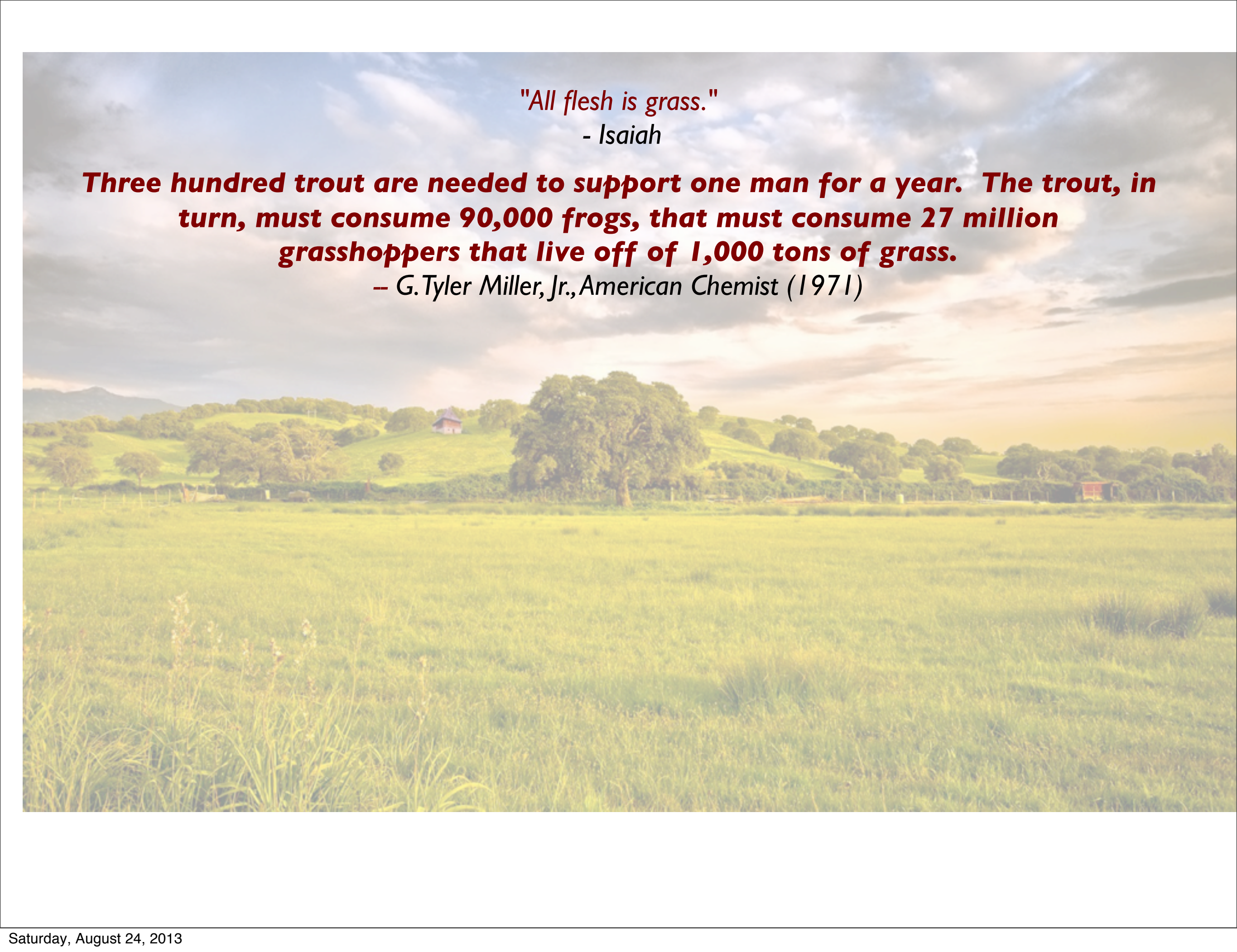
- **Trophic Efficiency**, the percent of production transferred one trophic level to the next. (Ranges between 5-20%)
- Trophic efficiency must be less than production efficiency.





Each tier represents the standing crop (the total dry mass of all organisms) in a trophic level.

Can you make an argument that humans might be better off if they were vegetarians?



*"All flesh is grass."
- Isaiah*

Three hundred trout are needed to support one man for a year. The trout, in turn, must consume 90,000 frogs, that must consume 27 million grasshoppers that live off of 1,000 tons of grass.

– G.Tyler Miller, Jr., American Chemist (1971)

Ecosystems

IV.

Main Idea: Most ecosystems receive an abundant supply of solar energy but chemical elements are they usually limited.

Main Idea: Solar energy is continually bombarding the earth but chemical elements on earth are finite and must be recycled.



BIOGEOCHEMICAL PROCESSES CYCLE NUTRIENTS AND WATER IN ECOSYSTEMS

- Nutrients are recycled using both *biotic and abiotic* processes together, hence the name *biogeochemical cycles*.

A. Biogeochemical Cycles

- These cycles occur on both on a local and global level.
- **A general look at cycles finds two key components:**
 - ***A Reservoir* (the location of the element)**
 - ***A Process* (the means of moving the element)**





Biogeochemical Cycles

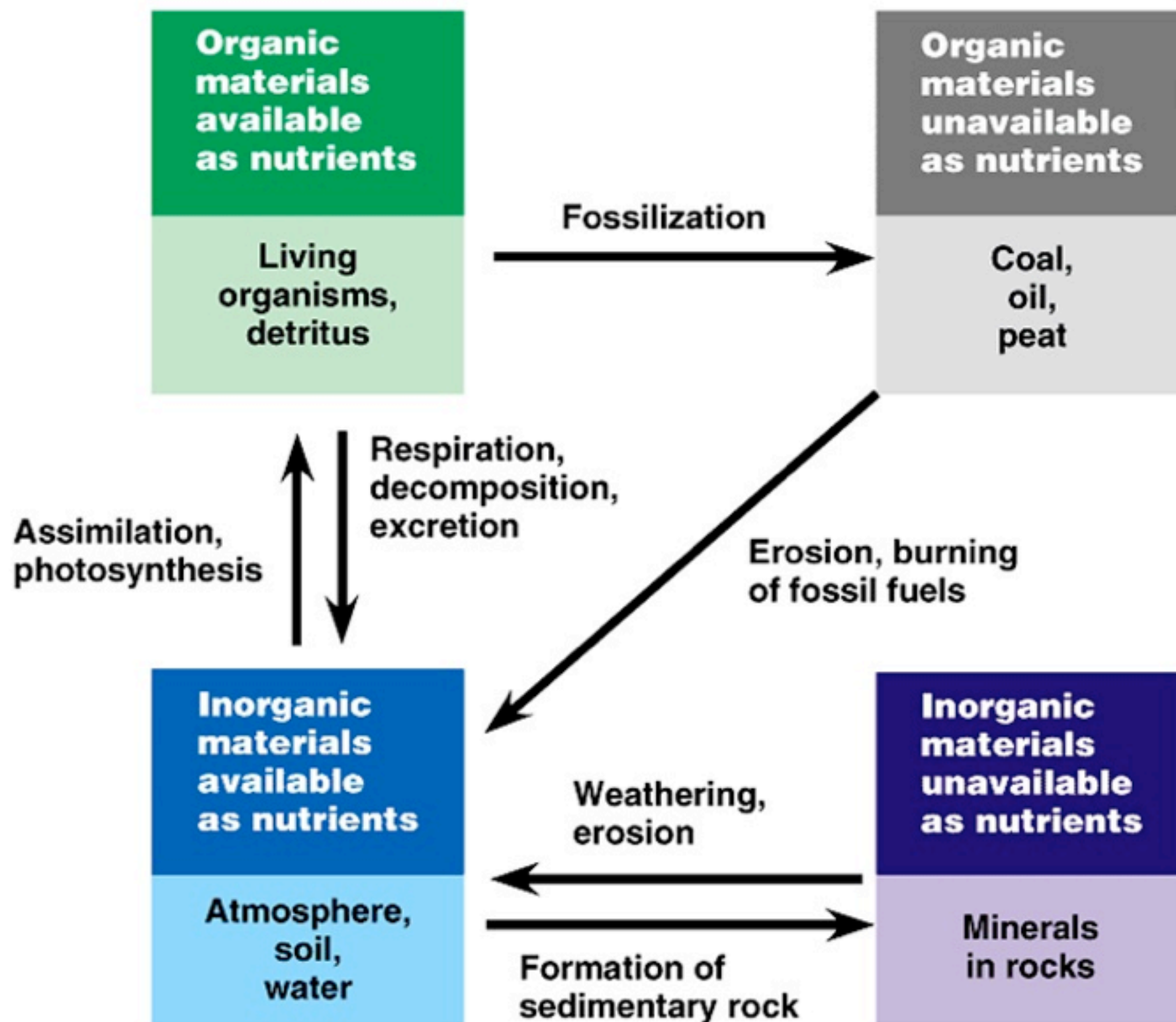
- Reservoirs.
 - Atmosphere
 - Hydrosphere
 - Lithosphere
 - Biosphere
- Chemicals in these reservoirs have different average storage times
 - Long in lithosphere (rocks)
 - Short in the atmosphere
 - Intermediate in the hydrosphere and biosphere

kinda cool periodic table...

1 H Hydrogen	<div>Atomic number → 20</div> <div>Environmentally important trace elements →</div> <div>Element relatively abundant in the Earth's crust</div> <div>Element symbol</div> <div>Ca</div> <div>□ □</div> <div>Calcium</div> <div>Name</div>																2 He Helium	
3 Li Lithium	4 Be Beryllium																	10 Ne Neon
11 Na Sodium	12 Mg Magnesium																	18 Ar Argon
19 K Potassium	20 Ca Calcium	21 Sc Scandium	22 Ti Titanium	23 V Vanadium	24 Cr Chromium	25 Mn Manganese	26 Fe Iron	27 Co Cobalt	28 Ni Nickel	29 Cu Copper	30 Zn Zinc	31 Ga Gallium	32 Ge Germanium	33 As Arsenic	34 Se Selenium	35 Br Bromine	36 Kr Krypton	
37 Rb Rubidium	38 Sr Strontium	39 Y Yttrium	40 Zr Zirconium	41 Nb Niobium	42 Mo Molybdenum	43 Tc Technetium	44 Ru Ruthenium	45 Rh Rhodium	46 Pd Palladium	47 Ag Silver	48 Cd Cadmium	49 In Indium	50 Sn Tin	51 Sb Antimony	52 Te Tellurium	53 I Iodine	54 Xe Xenon	
55 Cs Cesium	56 Ba Barium	57 La Lanthanum	72 Hf Hafnium	73 Ta Tantalum	74 W Tungsten	75 Re Rhenium	76 Os Osmium	77 Ir Iridium	78 Pt Platinum	79 Au Gold	80 Hg Mercury	81 Tl Thallium	82 Pb Lead	83 Bi Bismuth	84 Po Polonium	85 At Astatine	86 Rn Radon	
87 Fr Francium	88 Ra Radium	89 Ac Actinium	104 Rf Rutherfordium	105 Db Dubnium	106 Sg Seaborgium	107 Bh Bohrium	108 Hs Hassium	109 Mt Meitnerium										

58 Ce Cerium	59 Pr Praseodymium	60 Nd Neodymium	61 Pm Promethium	62 Sm Samarium	63 Eu Europium	64 Gd Gadolinium	65 Tb Terbium	66 Dy Dysprosium	67 Ho Holmium	68 Er Erbium	69 Tm Thulium	70 Yb Ytterbium	71 Lu Lutetium
90 Th Thorium	91 Pa Protactinium	92 U Uranium	93 Np Neptunium	94 Pu Plutonium	95 Am Americium	96 Cm Curium	97 Bk Berkelium	98 Cf Californium	99 Es Einsteinium	100 Fm Fermium	101 Md Mendelevium	102 No Nobelium	103 Lw Lawrencium

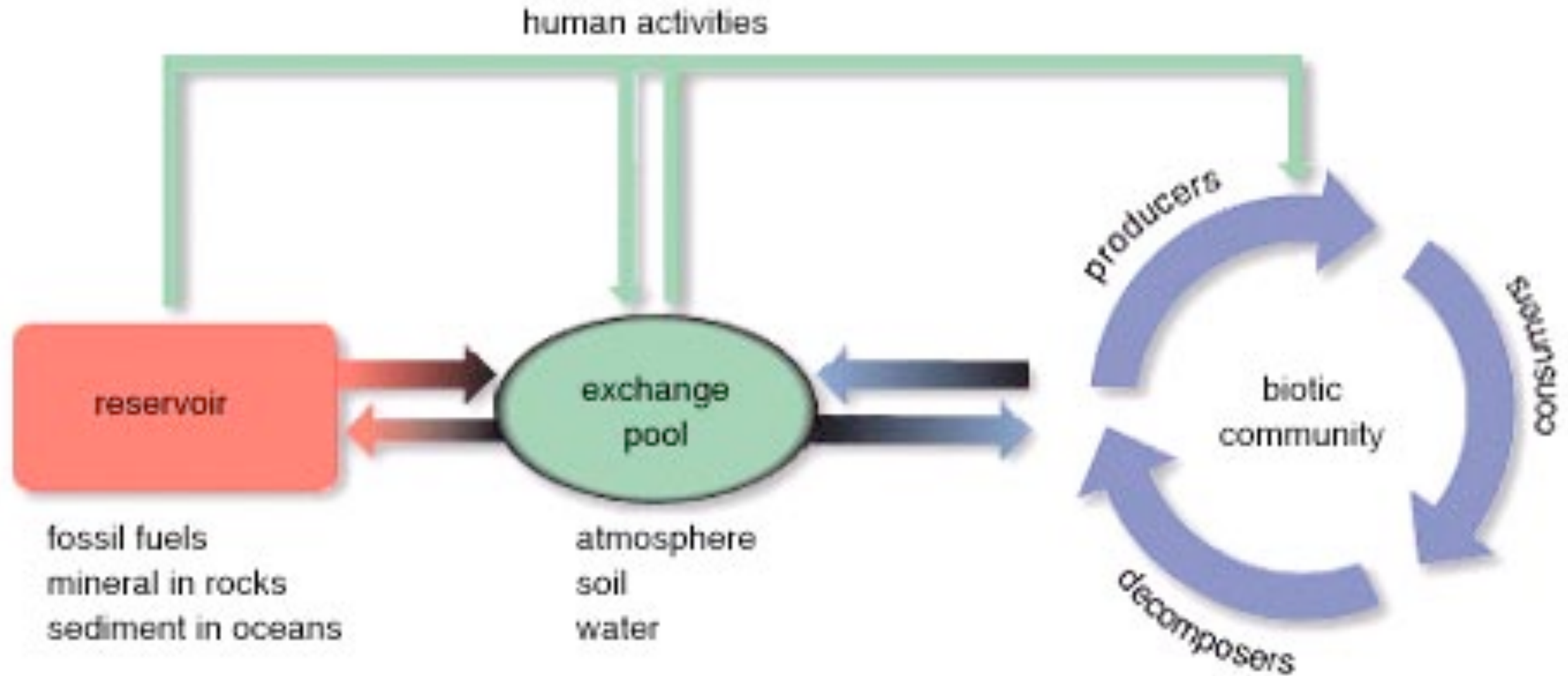
-  = Required for all life
-  = Required for some life-forms
-  = Moderately toxic: either slightly toxic to all life or highly toxic to a few forms
-  = Highly toxic to all organisms, even in low concentrations



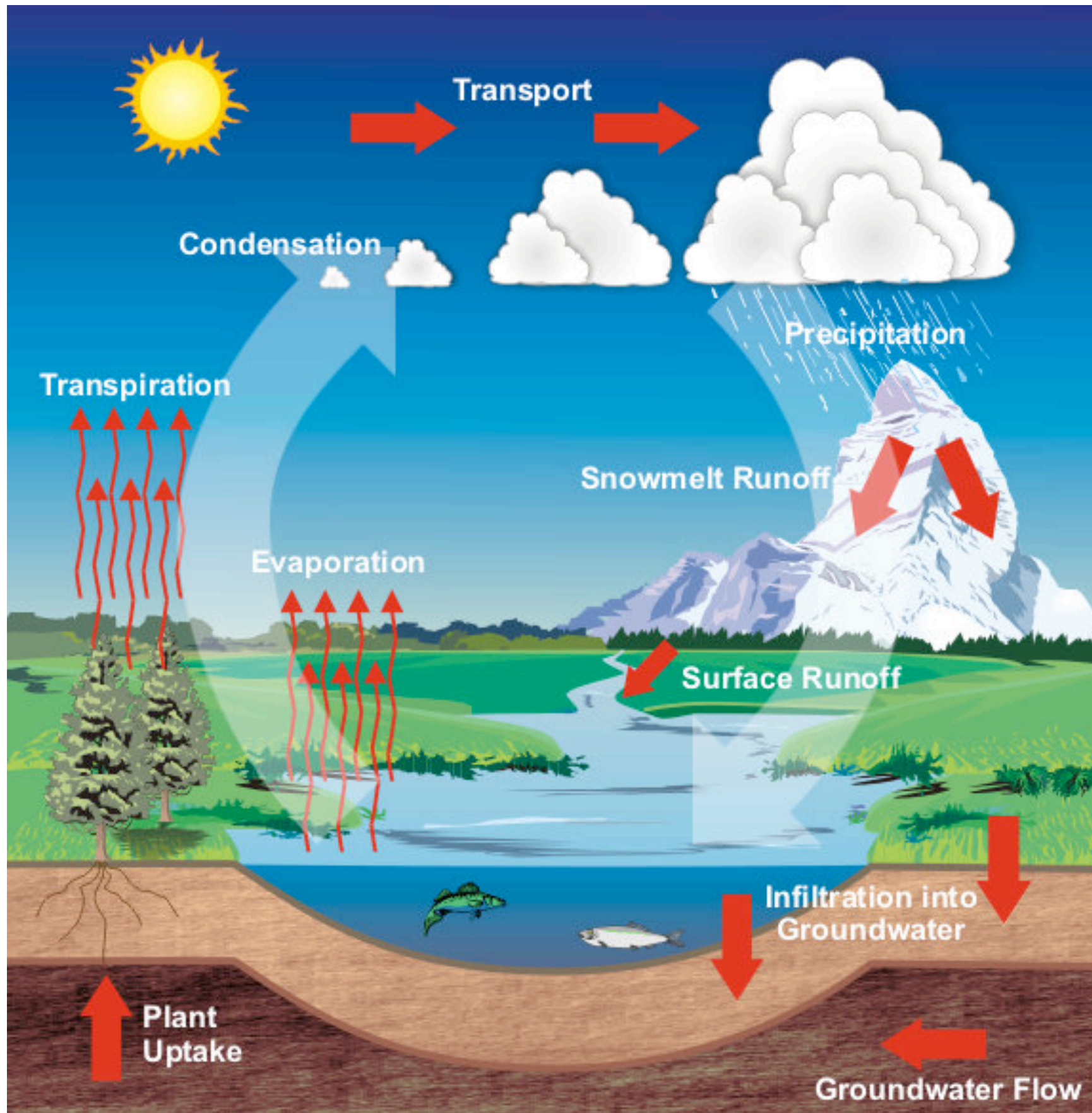
Can you match these reservoirs using the terms on the previous slide?

Simplified Version

A closer look!



WATER CYCLE



BIOLOGICAL IMPORTANCE

- Water is essential for all life, water also influences production & decomposition

FORMS AVAILABLE TO LIFE

- Mainly liquid water

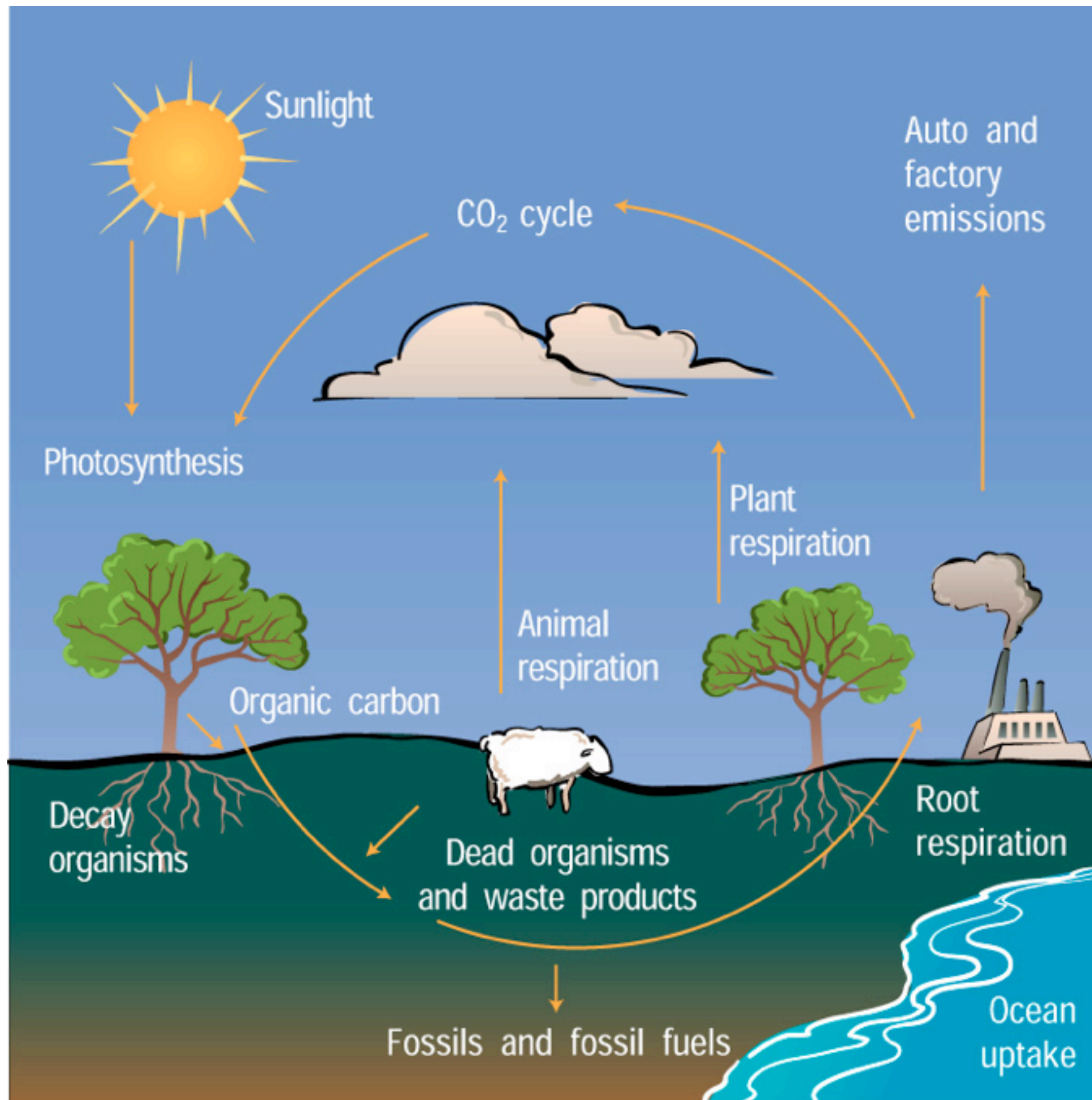
RESERVOIRS

- Rough estimations: 97% in oceans, 2% in glaciers and ice caps, 1% in rivers and lakes

KEY PROCESSES

- Evaporation, Condensation and Precipitation

CARBON CYCLE



BIOLOGICAL IMPORTANCE

- Carbon is the backbone of all organic compounds essential for life.

FORMS AVAILABLE TO LIFE

- CO₂ used by autotrophs, many other organic forms used by the rest of life.

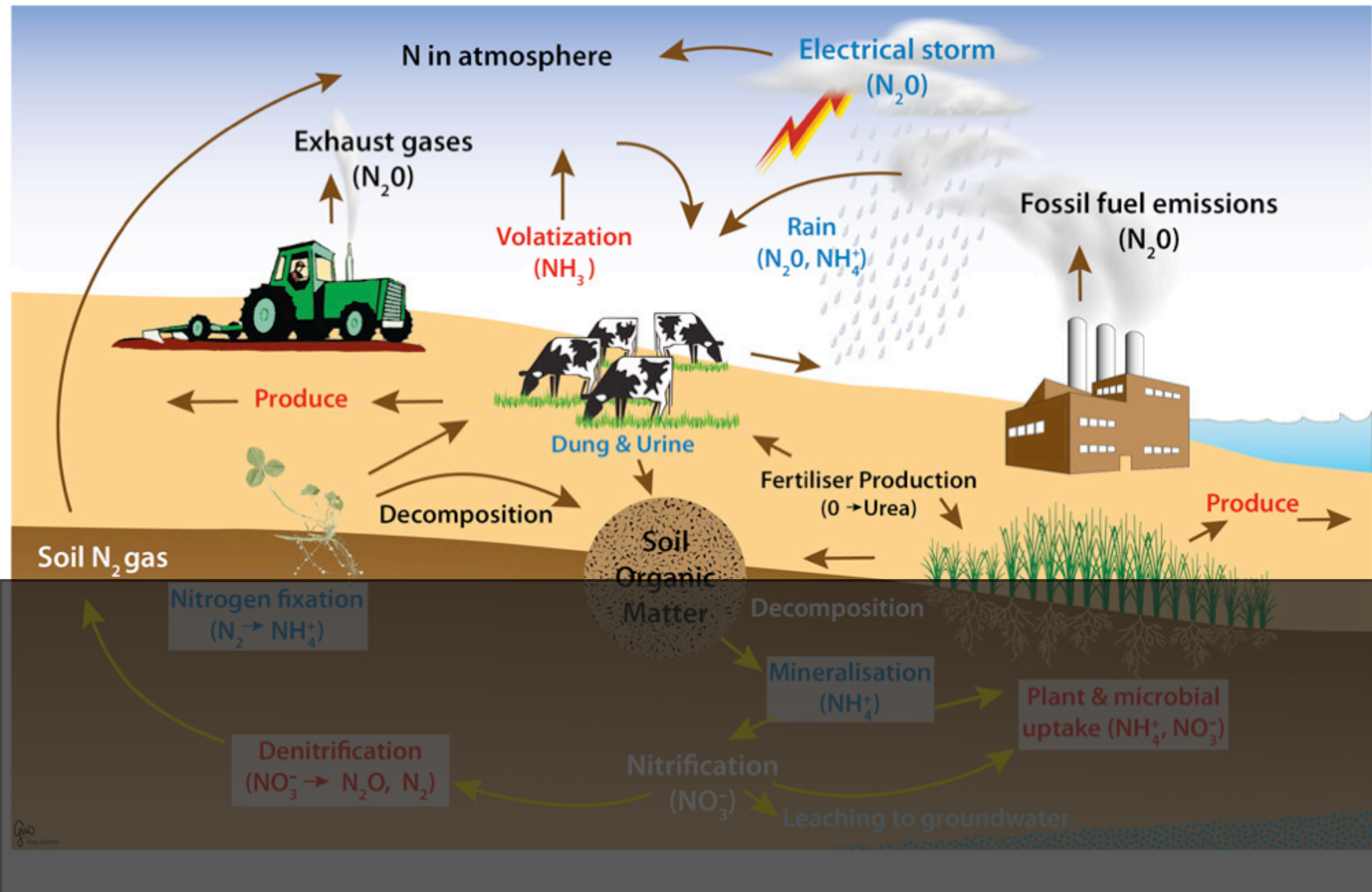
RESERVOIRS

- Fossil fuels, sediments of aquatic ecosystems, dissolved carbon in oceans, plant/animal biomass, atmosphere, sedimentary rocks (the largest)

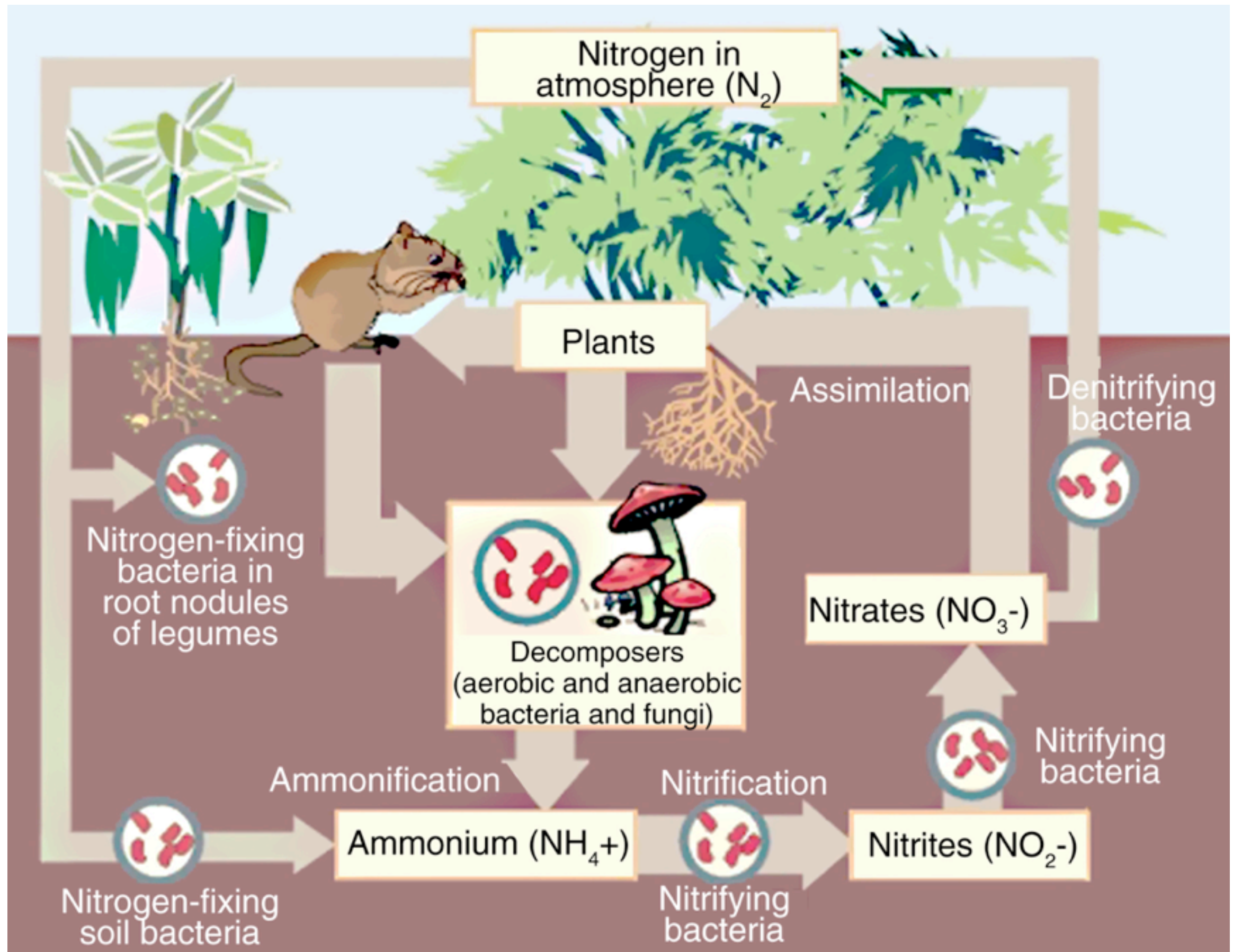
KEY PROCESSES

- Mainly photosynthesis and cellular respiration, burning of fossil fuels, volcanoes

NITROGEN CYCLE (abiotic +)



NITROGEN CYCLE (biotic focus)



BIOLOGICAL IMPORTANCE

- Nitrogen is an important part of proteins and nucleic acids.

FORMS AVAILABLE TO LIFE

- Bacteria can use ammonium (NH_4^+), nitrates (NO_3^-), nitrites (NO_2^-) and some organic forms. Plants use all of the above except nitrites (NO_2^-). Animals can only use organic forms.

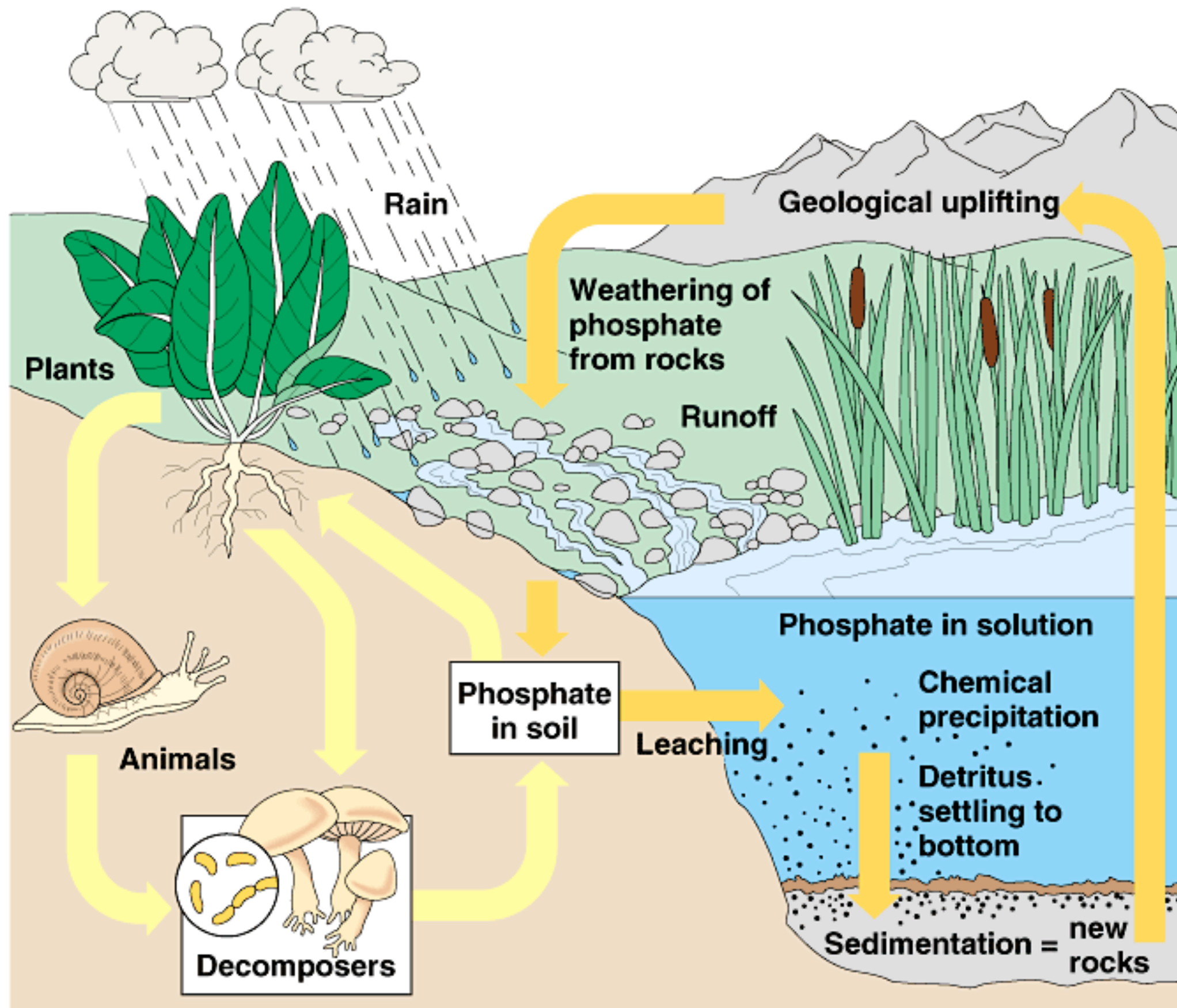
RESERVOIRS

- Atmosphere(the largest), soils, sediments of aquatic ecosystems, dissolved in water and biomass of living organisms

KEY PROCESSES

- Mainly nitrogen fixation, lightning, industrial fertilizers

PHOSPHORUS CYCLE



BIOLOGICAL IMPORTANCE

- Phosphorus is an important part of phospholipids (needed to make cell membranes), nucleic acids and ATP. In addition phosphorus is a mineral constituent of bones and teeth.

FORMS AVAILABLE TO LIFE

- Phosphates (PO_4^{3-}) absorbed by plants

RESERVOIRS

- Sedimentary rocks, soil, dissolved in the ocean and in biomass of organisms.

KEY PROCESSES

- Weathering of rocks, leaching from soil, eaten by consumers, excretion by organisms

B. Decomposition & Nutrient Cycle Rates

- Decomposition is **essential** for nutrient cycling.
- Decomposition is **highly variable**, mainly due to differences in rates of decomposition.
- Decomposition is once again **under the influence of**:
 - *Temperature, Moisture and Nutrient Availability*
 - There is an optimum temperature for decomposition

What type of curve would you predict if graphed temperature and rates of decomposition?

Which variable belongs on the X axis? Y axis?

Identify the biomes below.

Which ecosystem has better (more fertile) soil?



- Decomposition in Tropical Rain Forests occurs rapidly
- As a result nutrients spend little time in the soil
- Ironically there soil is low nutrients, about 10% of ecosystems total

- Decomposition in Temperate Forests occurs less rapidly
- As a result nutrients spend more time in the soil
- The soil has a moderate amount of nutrients, about 50% of ecosystems total